

**Haruko Takechi,^a Kanji Kubo,^{b*}
Hajime Takahashi^a and Taisuke
Matsumoto^c**

^aFaculty of Pharmaceutical Sciences, Health Sciences University of Hokkaido, 1757 Kanazawa, Ishikari-Tobetsu, Hokkaido 061-0293, Japan, ^bSchool of Dentistry, Health Sciences University of Hokkaido, 1757 Kanazawa, Ishikari-Tobetsu, Hokkaido 061-0293, Japan, and

^cInstitute for Materials Chemistry and Engineering, Kyushu University, Kasuga-koen, Kasuga, Fukuoka 816-8580, Japan

Correspondence e-mail:
kubo-k@hoku-iryo-u.ac.jp

Key indicators

Single-crystal X-ray study

$T = 123\text{ K}$

Mean $\sigma(\text{C}-\text{C}) = 0.002\text{ \AA}$

R factor = 0.028

wR factor = 0.089

Data-to-parameter ratio = 17.1

For details of how these key indicators were automatically derived from the article, see <http://journals.iucr.org/e>.

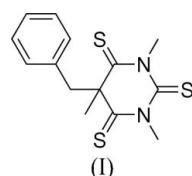
5-Benzyl-1,3,5-trimethylpyrimidine-2,4,6(1*H*,3*H*,5*H*)-trithione

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The rings in the title compound, $C_{14}H_{16}N_2S_3$, adopt a *syn* conformation with respect to each other. The pyrimidinetriithione ring system exhibits conjugation involving the lone pairs on the N atoms and the three C=S bonds. Intermolecular C–H···S interactions and S···S contacts help stabilize the crystal packing.

Comment

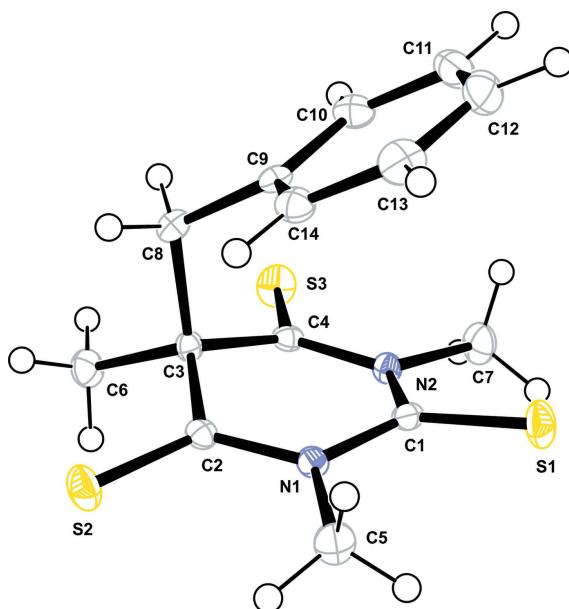
Barbiturates act as central nervous system depressants and produce a wide spectrum of effects, from mild sedation to anesthesia (Bideau *et al.*, 1976; Gotthardt *et al.*, 1980). Some are also used as anticonvulsants. Although a number of barbiturates have been studied, there have been only a few studies of their sulfur analogs (trithiobarbiturate) (Greenberg *et al.*, 2000) and while most crystals of barbiturates are colorless those of thiobarbiturates, such as (I), are usually colored.



In (I), shown in Fig. 1, the C–N bond lengths (Table 1) are shorter than the value of 1.416 Å (Allen *et al.*, 1987) usually observed for Csp^2-Nsp^3 bonds, indicating that the lone pairs on N1 and N2 are conjugated with the three C=S bonds. The heterocyclic six-membered ring has a flattened boat conformation. The pyrimidinetriithione group and the phenyl ring are in a *syn* conformation with respect to each other. There are some close intermolecular S···S contacts in (I); S1···S2ⁱ and S2···S1ⁱⁱ at 3.5076 (4) Å [symmetry codes: (i) $x, 1 + y, z$; (ii) $x, y - 1, z$] lie within the range of S···S contacts (3.35–3.65 Å) found in similar compounds (Dai *et al.*, 1997; Zhou *et al.*, 2001). Some intermolecular C–H···S interactions are also observed in (I) (Table 2). These S···S contacts and C–H···S interactions help stabilize the crystal packing (Fig. 2).

Experimental

Compound (I) was synthesized by using 1,3-dimethyl-2-thiourea and benzylmethylmalonyl chloride in triethylamine (Takechi *et al.*, 1999) to produce dihydro-1,3,5-trimethyl-5-(phenylmethyl)-2-thioxo-4,6(1*H*,3*H*)-pyrimidinedione, (II). Reflux of (II) (15 mmol) and Lawesson's reagent (9.7 g, 24 mmol) in xylene (40 ml) for 24 h afforded compound (I) in 24% yield. Crystals of (I) were grown from a chloroform solution by slow evaporation.

**Figure 1**

Molecular structure of (I), showing 50% probability displacement ellipsoids.

Crystal data

$C_{14}H_{16}N_2S_3$	$Z = 4$
$M_r = 308.47$	$D_x = 1.407 \text{ Mg m}^{-3}$
Monoclinic, $P2_1/a$	Mo $K\alpha$ radiation
$a = 13.112 (13) \text{ \AA}$	$\mu = 0.50 \text{ mm}^{-1}$
$b = 8.350 (8) \text{ \AA}$	$T = 123.1 \text{ K}$
$c = 13.299 (13) \text{ \AA}$	Prism, red
$\beta = 90.721 (7)^\circ$	$0.15 \times 0.13 \times 0.10 \text{ mm}$
$V = 1456 (2) \text{ \AA}^3$	

Data collection

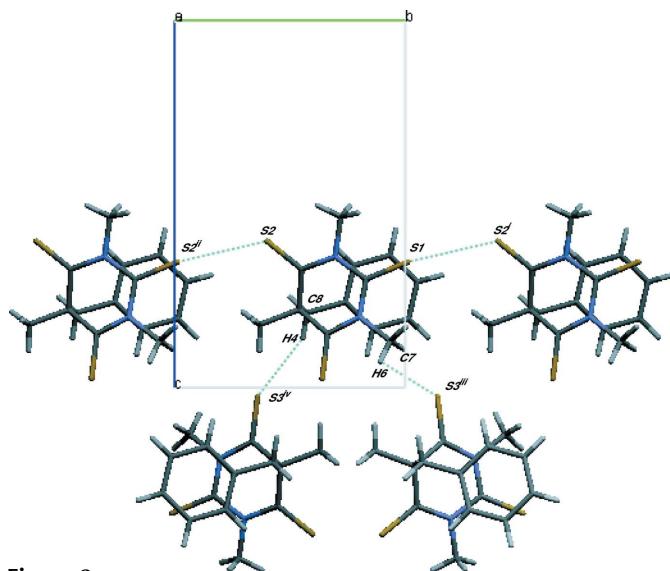
Rigaku Saturn diffractometer	16694 measured reflections
ω scans	3216 independent reflections
Absorption correction: multi-scan (REQAB; Jacobson, 1998)	2731 reflections with $F^2 > 2\sigma(F^2)$
$T_{\min} = 0.891$, $T_{\max} = 0.952$	$R_{\text{int}} = 0.035$
	$\theta_{\max} = 27.5^\circ$

Refinement

Refinement on F^2	H-atom parameters constrained
$R[F^2 > 2\sigma(F^2)] = 0.028$	$w = 1/[0.0007F_o^2 + \sigma(F_o^2)]/(4F_o^2)$
$wR(F^2) = 0.089$	$(\Delta/\sigma)_{\max} < 0.001$
$S = 1.06$	$\Delta\rho_{\max} = 0.40 \text{ e \AA}^{-3}$
3216 reflections	$\Delta\rho_{\min} = -0.26 \text{ e \AA}^{-3}$
188 parameters	

Table 1
Selected geometric parameters (\AA , $^\circ$).

S1–C1	1.6447 (13)	N2–C1	1.3941 (17)
S2–C2	1.6436 (13)	N2–C4	1.3658 (16)
S3–C4	1.6455 (13)	N2–C7	1.4745 (17)
N1–C1	1.3897 (16)	C2–C3	1.5162 (18)
N1–C2	1.3757 (16)	C3–C4	1.5182 (18)
N1–C5	1.4736 (17)		
C5–N1–C1–S1	−11.48 (16)	S2–C2–C3–C8	80.73 (12)
C5–N1–C2–S2	−2.78 (16)	C2–C3–C8–C9	64.59 (13)
C7–N2–C1–S1	10.58 (16)	C4–C3–C8–C9	−57.31 (14)
C7–N2–C4–S3	3.69 (16)	C6–C3–C8–C9	−176.19 (11)
S2–C2–C3–C6	−35.98 (15)		

**Figure 2**

Intermolecular C–H···S interactions and S···S contacts of (I). Broken lines indicate these interactions. [Symmetry codes: (i) $x, 1 + y, z$; (ii) $x, y - 1, z$; (iii) $\frac{1}{2} - x, \frac{1}{2} + y, 2 - z$; (iv) $2 - x, 1 - y, 2 - z$.]

Table 2

Hydrogen-bond geometry (\AA , $^\circ$).

$D-\text{H}\cdots A$	$D-\text{H}$	$\text{H}\cdots A$	$D\cdots A$	$D-\text{H}\cdots A$
$C7-\text{H}_6\cdots S3^{iii}$	0.95	2.95	3.415 (2)	111
$C8-\text{H}_4\cdots S3^{iv}$	0.95	2.88	3.749 (4)	153

Symmetry codes: (iii) $-x + \frac{3}{2}, y + \frac{1}{2}, -z + 2$; (iv) $-x + 2, -y + 1, -z + 2$.

H atoms bonded to C atoms were included in the refinement at calculated positions and refined with a riding model, with $C-\text{H} = 0.95 \text{ \AA}$ and $U_{\text{iso}}(\text{H}) = 1.2U_{\text{eq}}(\text{C})$.

Data collection: *CrystalClear* (Rigaku, 1999); cell refinement: *CrystalClear*; data reduction: *CrystalStructure* (Rigaku/MSC, 2004); program(s) used to solve structure: *SIR97* (Altomare *et al.*, 1999); program(s) used to refine structure: *CRYSTALS* (Betteridge *et al.*, 2003); molecular graphics: *ORTEPIII* (Burnett & Johnson, 1996) and *Mercury* (Version 1.3; Bruno *et al.*, 2002); software used to prepare material for publication: *CrystalStructure*.

References

- Allen, F. H., Kennard, O., Watson, D. G., Brammer, L., Orpen, A. G. & Taylor, R. (1987). *J. Chem. Soc. Perkin Trans. 2*, pp. S1–19.
- Altomare, A., Burla, M. C., Camalli, M., Cascarano, G. L., Giacovazzo, C., Guagliardi, A., Molterni, A. G. G., Polidori, G. & Spagna, R. (1999). *J. Appl. Cryst.* **32**, 115–119.
- Betteridge, P. W., Carruthers, J. R., Cooper, R. I., Prout, K. & Watkin, D. J. (2003). *J. Appl. Cryst.* **36**, 1487.
- Bideau, J. P., Huong, P. V. & Toure, S. (1976). *Acta Cryst.* **B32**, 481–488.
- Bruno, I. J., Cole, J. C., Edgington, P. R., Kessler, M., Macrae, C. F., McCabe, P., Pearson, J. & Taylor, R. (2002). *Acta Cryst.* **B58**, 389–397.
- Burnett, M. N. & Johnson, C. K. (1996). *ORTEPIII*. Report ORNL-6895. Ridge National Laboratory, Tennessee, USA.
- Dai, J., Munakata, M., Wu, L. P., Kuroda-Sowa, T. & Suenaga, Y. (1997). *Inorg. Chim. Acta*, **258**, 65–69.
- Gotthardt, H., Niebel, S. & Doenecke, J. (1980). *Liebigs Ann. Chem.* pp. 873–885.
- Greenberg, M., Shtelman, V. & Kaftory, M. (2000). *Acta Cryst.* **C56**, 465–468.
- Jacobson, R. (1998). *REQAB*. Private communication to the Rigaku Corporation, Tokyo, Japan.

Rigaku (1999). *CrystalClear*. Rigaku Corporation, Tokyo, Japan.
Rigaku/MSC (2004). *CrystalStructure*. Version 3.7.0. Rigaku/MSC, The Woodlands, Texas, USA.

Takechi, H., Takahashi, H. & Machida, M. (1999). *Heterocycles*, **50**, 159–168.
Zhou, Z.-R., Xu, W., Xia, Y., Wang, Q.-R., Ding, Z.-B., Chen, M.-Q., Hua, Z.-Y. & Tao, F.-G. (2001). *Acta Cryst. C* **57**, 471–472.